

FORM PTO-1449 (Rev. 2-32) U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE INFORMATION DISCLOSURE STATEMENT BY APPLICANT (Use several sheets if necessary)	ATTY. DOCKET NO. RE26499A	SERIAL NO.
	APPLICANT: KLABUNDE, Kenneth J. et al.	
	FILING DATE:	GROUP: 1752 1754

U.S. PATENT DOCUMENTS

EXAM. INITIAL	DOCUMENT NUMBER	DATE	NAME	CLASS	SUBCLASS	FILING DATE IF APPROPRIATE
CNN	2 4 7 4 2 0 7	06/28/1949	Lovell et al.	—	—	
CNN	4 6 3 6 3 7 8	01/13/1987	Bestek et al.	—	—	
CNN	6 0 6 0 2 8 1	06/27/2000	Attia	—	—	
CNN	5 8 5 8 2 1 2	01/12/1999	Darcy	—	—	
CNN	5 4 8 2 5 3 6	01/09/1996	Ngai et al.	—	—	
CNN	5 5 4 0 8 9 6	07/30/1996	Newby	—	—	
CNN	4 3 2 4 7 7 6	04/13/1982	Kim	—	—	
CNN	4 0 0 2 7 2 0	01/11/1997	Wheelock et al.	—	—	
CNN	4 0 4 5 3 7 1	08/30/1977	Wheelock et al.	—	—	
CNN	5 1 7 3 2 7 9	12/22/1992	Dumont et al.	—	—	

FOREIGN PATENT DOCUMENTS

DOCUMENT NUMBER	DATE	COUNTRY	CLASS	SUBCLASS	TRANSLATION
					YES NO

OTHER DOCUMENTS (Including Author, Title, Date, Pertinent Pages, Etc.)

CNN		Effect of Ferric Oxide on the High-Temperature Removal of Hydrogen Sulfide over ZnO-Fe ₂ O ₃ Mixed Metal Oxide Sorbent; Young-Soo Lee, Hee-Taik Kim, and Kyong-Ok Yoo; <u>Ind. Eng. Chem. Res.</u> , Vol. 34, No. 4, 1995; pgs. 1181-1188
CNN		Characterization of Reaction between Zinc Oxide and Hydrogen Sulfide; Eiji Sasaoka; <u>Energy & Fuels</u> , Vol. 8, No. 5, 1994; pgs. 1100-1105
CNN		Kinetics of the Absorption of Hydrogen Sulfide by High Purity and Doped High Surface Area Zinc Oxide; J. Michael Davidson, Catriona H. Lawrie and Khalid Sohail; <u>Ind. Eng. Chem. Res.</u> , Vol. 34, No. 9, 1995; pgs. 2981-2989
CNN		A DRIFTS Study of the Surface and Bulk Reactions of Hydrogen Sulfide with High Surface Area Zinc Oxide; J. Michael Davidson and Khalid Sohail; <u>Ind. Eng. Chem. Res.</u> , Vol. 34, No. 11, 1995; pgs. 3675-3677

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FORM PTO-1449 (Rev. 2-32) U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE INFORMATION DISCLOSURE STATEMENT BY APPLICANT (Use several sheets if necessary)	ATTY. DOCKET NO. RE26499A	SERIAL NO. 09/933,474
	APPLICANT: KLABUNDE, KENNETH et al.	
	FILING DATE: August 17, 2001	GROUP: 1754

U.S. PATENT DOCUMENTS

EXAM. INITIAL	DOCUMENT NUMBER	DATE	NAME	CLASS	SUBCLASS	FILING DATE IF APPROPRIATE
Can	4 3 1 4 9 3 2	2/1982	Wakimoto et al.			
	4 5 0 8 8 4 1	4/1985	Onuma et al.			
	4 5 4 3 3 4 1	9/1985	Barringer et al.			
	4 7 5 5 3 6 5	7/1988	Funahashi et al.			
	5 0 0 8 2 2 1	4/1991	Ketcham			
	5 2 9 0 3 3 2	3/1994	Chatterjee et al.			
	5 3 5 8 9 1 3	10/1994	Chatterjee et al.			
	5 4 2 0 0 8 6	5/1995	Brandau et al.			
	5 4 6 3 1 6 7	10/1995	Ou			
	5 5 4 0 9 8 1	7/1996	Gallagher et al.			
	5 6 7 0 2 4 7	9/1997	Takaoka et al.			
	5 7 1 2 2 1 9	1/1998	Klabunde et al.			
	5 7 5 9 9 3 9	6/1998	Klabunde et al.			
	5 8 0 7 7 9 8	9/1998	Bolt et al.			
Can	5 9 1 4 4 3 6	6/1999	Klabunde et al.			

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DOCUMENT NUMBER	DATE	COUNTRY	CLASS	SUBCLASS	TRANSLATION
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OTHER DOCUMENTS (Including Author, Title, Date, Pertinent Pages, Etc.)

Can	Ganguly et al.; Sol-Gel Microsphere Pelletization Process for Fabrication of High-Density ThO ₂ -2% UO ₂ Fuel for Advanced Pressurized Heavy Water Reactors; <i>Nuclear Technology</i> , Vol. 73: pp. 84-94, Apr. 1986
Can	Cano et al.; Development of direct reduction pellets containing MgO by Samarco Mineracao S/A; <i>Mining Engineering</i> ; June 1993, pp. 633-636
Can	Browning; Agglomeration: Growing Larger in Applications and Technology; <i>Chemical Engineering</i> ; Dec. 4, 1967, pp. 147-169
Can	Agglomeration; <i>Chemical Engineering</i> ; Vol. 58, No. 10, Oct. 1951, pp. 161-170

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